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Chemistry Chapter 4 and 5 vocab. Atomic number. Mass number. Isotope. Atomic mass unit (amu) The number of protons in the nucleus of an atom of an element. Total

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number of protons and neutrons in the nucleus of an atom. Atoms of the same element that have same atomic number but dif.... Unit of mass equal to one-twelfth the mass of a carbon-12 atom.

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Chapter 4, Lesson 5:  
Energy Levels,  
Electrons, and Ionic

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Bonding. Key Concepts. • The attractions between the protons and electrons of atoms can cause an electron to move completely from one atom to the other. • When an atom loses or gains an electron, it is called an ion.

## **Chapter 4, Lesson 5: Energy Levels, Electrons, and Ionic**

...

Mrs. Gingras'  
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Chemistry Page. Home  
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Chemistry 122 About  
Contact Chapters 4, 5  
& 6. This unit's focus is  
on atomic structure. ...

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for Consumers.

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Understanding Rate of  
Reaction; 02 Analysing  
Rate of Reaction from  
Graph; 03 Factors

Affecting the Rate of  
Reaction; 04 Collision  
Theory; 05 Applications  
of Rate of Reaction;

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Compound; 02 Alkane;  
03 Structural Formula  
of Carbon Compounds;

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Stoichiometry of  
Chemical Reactions.

Introduction; 4.1

Writing and Balancing  
Chemical Equations;

4.2 Classifying

Chemical Reactions;

4.3 Reaction

Stoichiometry; 4.4

Reaction Yields; 4.5

Quantitative Chemical

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4.3 Heat of

Displacement 4.4 Heat

of Neutralisation 4.5

Heat of Combustion.

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Electrons; Lesson 2:  
The Periodic Table;  
Lesson 3: The Periodic  
Table & Energy Level  
Models; Lesson 4:  
Energy Levels,  
Electrons, and  
Covalent Bonding;  
Lesson 5: Energy  
Levels, Electrons, and

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Ionic Bonding; Lesson  
6: Represent Bonding  
with Lewis Dot  
Diagrams; Chapter 5.  
Lesson 1: Water is a  
Polar ...

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Lewis dot structures  
provide a simple model  
for rationalizing the  
bonding in most known  
compounds. However,  
there are four general  
exceptions to the octet

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rule: (1) molecules, such as NO, with an odd number of electrons; (2) molecules in which one or more atoms possess more than eight electrons, such as SF<sub>6</sub>; (3) molecules such as BCl<sub>3</sub>, in which one or more atoms possess less than eight ...

**Chapter 4.5:  
Exceptions to the  
Octet Rule -  
Chemistry ...**

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Revision Notes Redox  
Reaction Redox  
Reaction in Terms of  
Gain and Loss of  
Oxygen Redox Reaction  
in Terms of Gain and  
Loss of  
Hydrogen Oxidation  
Number Oxidation  
Number and IUPAC  
Nomenclature Redox  
Reaction in Terms of  
Change of Oxidation  
Number Redox Reaction  
in Terms of Transfer of  
Electron Types of Redox  
Reaction Oxidising and

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Reducing  
Agents Change of  
Iron(II) to Iron(III) and  
Vice  
Versa Displacement ...

## **SPM Form 5 Chemistry Chapter 3 - Oxidation and Reduction ...**

Introduction; 18.1  
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Occurrence and  
Preparation of the  
Representative Metals;  
18.3 Structure and  
General Properties of



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the Metalloids; 18.4  
Structure and General  
Properties of the  
Nonmetals; 18.5  
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Preparation, and  
Compounds of  
Hydrogen; 18.6  
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Preparation, and  
Properties of  
Carbonates; 18.7  
Occurrence,  
Preparation, and  
Properties of Nitrogen

**Answer Key Chapter**

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a quirky cat-girl, and  
her friends! Trying to  
maneuver her way  
through high school is  
hard enough as it is,  
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ideas, crazy crushes,  
and some straight up  
CA-razy villains add to  
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